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(54) Wrapper for smoking articles and method for preparing same.

(57) A wrapper 102 for a cigarette 100 having controllable combustion characteristics including reduced visible sidestream smoke. The wrapper comprises cigarette paper and as an additive at least one of aluminium oxide, zirconium oxide, titanium oxide, calcium oxide, magnesium aluminate and calcium aluminate applied in an annular band 101 to the cigarette paper.

★ PHIM - Fee 93-282148/98 ★ EP 559300-A2  
Wrapper for smoking articles - comprises paper fibres and additive of one or more of alumina, stroma, titania, calcia, magnesium aluminate or calcium aluminate (Eag)

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D18 E37 P15 (93.09.08) A24D 1/02  
90.02.06 93EP-201332 Related to 90.02.06 90EP-301216 R(AT BE CH DE DK ES FR GB GR IT LI LU NL SE)  
Wrapper for smoking articles comprises paper fibres (I), together with an additive (II). (II) is 1 or more of Al<sub>2</sub>O<sub>3</sub>, ZrO<sub>2</sub>, TiO<sub>2</sub>, CaO, Mg aluminate and Ca aluminate.

ADVANTAGE - The wrapper contg. the specific inorganic additive(s) (II) has controllable combustion characteristics, including reduced visible sidestream smoke, and a controlled mass burn rate of cigarettes. Smoking articles using the wrappers have the look, feel, and taste of conventional articles. In addn. the wrappers may be used in the prepn. of self-extinguishing cigarettes.  
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The present invention relates to the manufacture of wrappers for smoking articles such as cigarettes. More particularly, this invention relates to the use of gels produced by a solution-gelation or "sol-gel" process for controlling the combustion of wrappers for smoking articles. In one embodiment, the gels made by this process are applied as coatings to paper fibers before the paper is formed into wrappers for smoking articles. The coated paper wrappers of this invention are useful in reducing visible sidestream smoke and in reducing or preventing combustion of the wrappers of smoking articles. The wrappers are also useful in controlling the mass burn rate of cigarettes.

One problem with smoking articles such as cigarettes is the amount of sidestream smoke generated during burning. Sidestream smoke is the smoke generated by cigarettes when they are burning but not being drawn upon by the smoker, as when a cigarette is placed in an ashtray between puffs. Sidestream smoke may be objectionable to non-smokers.

Efforts have been made to reduce the amount of sidestream smoke generated by cigarettes. While these methods may reduce the amount of sidestream smoke, they do not provide the smoker with a cigarette that looks, feels, and tastes the same as conventional cigarettes.

U.S. Patent No. 4,231,377 to Cline et al. ("Cline I") refers to a method of reducing sidestream smoke by applying a specified amount of magnesium oxide or magnesium hydroxide in combination with specific chemical "adjuvants" to paper fibers as a filler or coating. The magnesium salts are applied as a solution. However, according to U.S. Patent No. 4,433,697 to Cline et al. ("Cline II"), the ash generated by burning cigarettes made from paper prepared according to the method of Cline I is unacceptable because of its flakiness. Cline II proposes to reduce the flakiness of the ash by adding ceramic fibers to the paper-making slurry. The method of Cline II is problematic because of the substantial cost of the ceramic fibers. A further shortcoming of the proposal of Cline II is that ceramic fibers must be added to the paper-making slurry; commercially available cigarette papers cannot be modified according to the proposal after their manufacture.

U.S. Patent No. 4,450,847 to Owens refers to the use of fine particle size magnesium hydroxide gel in place of the magnesium hydroxide referred to in Cline I and II. Although Owens characterizes his magnesium hydroxide as a gel, he states that the gel is actually "an apparently homogeneous substance or a dispersion consisting of an aggregate of very small particles in very close association with a liquid, and the gel at the concentrations used in this invention is actually broken into flocs

floating in the aqueous medium." (Col. 3, lines 62-68). Owens' magnesium hydroxide gel is said to be used with magnesium oxide or calcium carbonate and a suitable chemical "adjuvant" as in Cline I and II.

Hampl et al., U.K. Patent Application GB 2 191 930 A, refers to cigarette papers that have been formed with high superficial surface area fillers. These cigarette papers are stated to yield reduced sidestream smoke when used as wrappers for cigarettes. Fillers such as attapulgite clay, fumed or activated alumina, chalk, fumed silica, and peroxides of magnesium, calcium, and strontium are said to be useful. Other fillers such as the carbonates, phosphates, sulphates, aluminates, and silicates of certain metals are also said to be useful as long as their superficial surface area exceeds 20 m<sup>2</sup>/g. Hampl et al.'s method also requires the use of potassium or sodium salts of various acids to maintain the integrity of the cigarette ash.

Apart from the problems already noted, the modified wrappers of Cline I and II, Owens, and Hampl et al. all suffer from a serious disadvantage the smoking of cigarettes made from paper containing magnesium oxide or magnesium hydroxide produces an off taste. It is believed that this taste arises from the strongly alkaline nature of the magnesium and from the other chemicals used.

In addition, the size of the particles employed by Cline I and II, Owens, or Hampl is critical. If the particles are too small, they will be dislodged from the wrapper, creating an objectionable dust. If they are too large, they will obstruct the porosity of the wrapper, interfering with the combustion of the tobacco fuel rod.

Accordingly it is an object of this invention to provide a wrapper for smoking articles which produces substantially reduced amounts of sidestream smoke, which prevents reduces or prevents combustion of the wrappers of smoking articles, and which may be incorporated into a smoking article which has the look, feel, and taste of conventional smoking articles.

The present invention solves the problems associated with prior art wrappers for smoking articles by providing a wrapper with desirable combustion characteristics, yet which may be incorporated into smoking articles that have the look, feel, and taste of conventional cigarettes. According to the invention there is provided a wrapper for a smoking article comprising paper fibres and, as an additive, at least one of aluminium oxide, zirconium oxide, titanium oxide, calcium oxide, magnesium aluminate and calcium aluminate. The invention also encompasses a smoking article having a wrapper according to the invention.

The additive may be produced by a sol-gel process, which provides a film of metal oxide which

it passes through t  
flows through the f  
a flavored aerosol  
to the mouth end.

The following e  
ods of preparation

These example  
use described shou  
of, rather than limita

#### EXAMPLE I

##### Preparation Of Alumina Acid

Approximately  
sec-butoxide was m  
containing 50 ml of  
The mixture was di  
and each portion w  
stirring using a magi  
sufficient to produce  
mixture, was maint  
and for an additional  
a solution containing  
dissolved in 20 ml of  
one portion of the m  
solution was added  
hour. After an additi  
mixtures was stopper  
combined and this c  
to a temperature of 8

#### EXAMPLE II

##### Coating Papers With Gel Process

Commercial grade  
cigarette papers (TOD  
a division of P.H. G  
Pisgah Forest, NC 28  
of Example I. The gel  
perature of 80 °C thro  
Hand sheets of cigar  
the gel and the exce  
from the paper. The p  
wave oven (Sharp Car  
at medium low setting  
were then hand-press  
wrinkles.

can be applied as a coating or a cigarette paper.

A sol-gel process is a soft chemical method of  
preparative solid state chemistry. The product of  
the process is an amorphous inorganic network.  
See generally R. Roy, "Ceramics By the Solution-  
Sol-Gel-Route," *Science*, 238, pp. 1664-69 (1987)  
and H. Schmidt, "Chemistry of Material Preparation  
by The Sol-Gel Process," *J. Non-Crystalline Solids*,  
100, pp. 51-64 (1988).

A sol-gel process can be used to produce a  
gel of selected metal oxides that is applied as a  
film upon the surface of fibers of conventional  
cigarette paper.

At least one annular band of a gel formed by a  
sol-gel process can be applied to the surface of a  
cigarette in order to produce a cigarette whose  
mass burn rate will approach zero when allowed to  
idle for a predetermined amount of time. Prefer-  
ably, a plurality of discrete bands, each of which  
substantially circumscribes the periphery of a  
smoking article, are present. And in a yet further  
embodiment, the wrapper of a novel smoking arti-  
cle described in copending European Patent Ap-  
plication 0 352 109, published 24th January 1990,  
hereinafter identified as 'a smoking article of the  
type specified', is modified with the gels made as  
described herein in order to prevent combustion of  
the wrapper of this novel smoking article.

The invention will be further described by way  
of example with reference to the drawings, in which  
like reference characters refer to like parts through-  
out, and in which:

Figure 1 is a longitudinal cross-sectional view of  
a smoking article of the type specified;

Figure 2 is a perspective view of a cigarette  
according to the present invention;

Figure 3 is a radial cross-section view of the  
cigarette of Figure 2, taken from line 3-3 of  
Figure 2; and

Figures 4 and 5 are graphs of the mass burn  
rates of cigarettes according to the present in-  
vention.

The sol-gel process permits synthesis of  
homogeneous gels of metal oxides. When dried  
after application to wrappers for smoking articles,  
these gels form films that reduce the amount of  
sidestream smoke produced during smoking. Addi-  
tionally, because dried films produced by a sol-gel  
process reduce the propensity of smoking articles  
to burn, they may be employed to produce self-  
extinguishing cigarettes and to substantially elimi-  
nate flaming combustion of novel smoking articles.

The sol-gel process generally consists of the  
following steps: preparation of a solution of a pre-  
cursor of a metal oxide, hydrolysis, condensation

or gelling, form-  
"Ceramics By The  
ence, 238, pp. 166  
of the gel resultin  
determined by the  
tion. Thus, the spe  
of the sol-gel proc  
ing the concentrati  
and pH during th  
steps.

The process f  
the present invent  
suitable metal oxi  
these substances  
to alkoxides, other  
rate and silicate  
inorganic salts or  
metal oxide precu

Water and a  
metal alkoxide, f  
and preferably a  
hydrolysis while  
may also be initi  
although the use  
may an organic  
acids, including h  
are preferred. Hy  
ferred. Organic ac  
citric acids are al  
tion. Increasing  
creasing the tem  
actions to occur  
are preferred, a  
control.

After hydroly  
continually stirred to  
of an acid or a n  
tion and affects  
The acid may b  
hydrolysis step.  
acetate, potassi  
sodium phosphat  
concentration of m  
viscosity of abou  
tion of the gelat  
have a metal o:  
18% with attend

After the de  
curred, the mixtu  
gel, in diluted or  
to paper fibers f  
may be applied  
known in the art.  
wrappers using  
blade coater, alt

\* In commercial embodiments of this invention, it is preferred to conduct this p  
atmosphere, such as under a blanket or stream of nitrogen.

ml (0.03 moles) of 1 N hydrochloric acid. The mixture was then heated with constant stirring as in Example I until complete solution was obtained. The mixture was then divided into two equal portions. Each portion was placed on a magnetic stirring hot plate and brought rapidly to a temperature of 95°C. Constant stirring and temperature were maintained for an additional 1.5 hours. After this time, 60 ml (0.06 moles) of 1 N hydrochloric acid was added to each portion. The temperature of the solution was maintained at 95°C and stirring continued for an additional 1.5 hours. The two mixtures were then combined and this mixture was brought to a temperature of 80°C.

#### EXAMPLE VI

##### Evaluation Of Wrappers Treated With Aluminum Oxide Gels

A film of the gel of Example V was deposited on the fibers of commercial grade calcium carbonate filled cigarette papers as described in Examples II and III and smoking articles of the type specified were prepared from them. The tests described in Example IV were performed on the smoking articles. In all three test situations, whether gels were applied by dipping (as in Example II) or with a size press (as in Example III) paper wrappers treated with the gel prepared by Example V resisted flaming combustion. No sidestream visible smoke was observed from the smoking articles prepared with papers treated by dipping, while slight sidestream smoke was observed from the smoking articles made with papers to which gel had been applied with a size press.

#### EXAMPLE VII

##### Preparation Of Dilute Aluminum Oxide Gels And Application To Cigarette Paper

In a nitrogen atmosphere, approximately 52 g (0.2 moles) of aluminum tri-sec-butoxide was mixed with 600 ml of distilled water containing 6 ml (0.06 moles) of 1 N hydrochloric acid. The solution was placed on a magnetic stirring hot plate, and the solution was rapidly brought to a boil with constant stirring as in Example I. Boiling and stirring continued for approximately 1 hour. Then 32 ml of 1 N hydrochloric acid (0.032 moles) was added to the solution. Boiling and stirring continued for an additional 5 minutes. At this time, the gel was divided into two equal portions.

One portion of the gel was retained as a source for concentrated gel. The other portion of the gel prepared from this procedure was diluted in a 50:50 ratio with distilled water. The gel of this

Example was applied to cigarette paper as described in Example II.

A second dilution of the gel was prepared by adding 50 ml of concentrated gel to 15 ml of distilled water. The same paper sample was dipped into this diluted gel and dried as in Example II. Finally, the same paper sample was dipped in the concentrated gel solution and dried as before. Then paper was the hand-pressed with an iron to remove wrinkles.

#### EXAMPLE VIII

##### Reduction Of Sidestream Smoke Production By Conventional Cigarettes

Cigarette paper (TOD 04242 and TOD 04244 obtained from Ecusta) was treated with the gels of Example VII at a variety of dilutions (10:1, 5:1, 2.5:1, 1:1, and undiluted). A film of gel at each dilution was applied to the cigarette paper by the method of Example II. Conventional cigarettes were prepared using these wrappers by hand wrapping a rod of commercial blend tobacco. These cigarettes were smoked on a single port smoking machine and the sidestream smoke production was evaluated by comparing light extinction with that observed when cigarettes prepared using untreated paper were smoked under the same conditions. Of those cigarettes that remained lit during the test, cigarettes made with sol gel treated wrappers produced 21% to 36% less sidestream smoke than did untreated cigarettes.

#### EXAMPLE IX

##### Preparation Of Magnesium Aluminate Gels And Treatment Of Cellulose Fibers

Approximately 120 g (0.5 moles) of aluminum sec-butoxide was mixed with 1.5 L of distilled water containing 15 ml (0.015 moles) of 1 N hydrochloric acid. The mixture was brought to a boil and continuously stirred as in Example I. Boiling and stirring continued until the mixture was reduced to a volume of approximately 870 ml. The mixture was then divided into one aliquot of 150 ml and six aliquots of 120 ml each.

One of the 120 ml aliquots was heated to a boil and then 3.8 g (0.033 moles) of magnesium ethoxide was added with stirring. After complete dissolution of the magnesium ethoxide, sufficient distilled water was added to bring the volume of the solution to 350 ml and then 12 ml (0.012 moles) of 1 N hydrochloric acid was added. The pH of this solution was about 12 and considerable foaming was present. Concentrated hydrochloric acid (approximately 12 N) was added dropwise until the

pH reached about 4. The volume of concentrated hydrochloric acid added was approximately 8 ml.

The gel of this Example was applied to commercial cigarette paper according to Example II and formed into a hollow tube for testing of its burn characteristics. The cigarette paper coated with the gel prepared as described in this Example resisted combustion when ignition with an open flame was attempted.

#### EXAMPLE X

##### Preparation Of Calcium Aluminate Gel

A solution of calcium ethoxide was prepared by reacting metallic calcium with absolute ethanol while heating the solution in a nitrogen atmosphere. A sample of 24.6 g of aluminum-sec-butoxide (0.1 moles) was mixed with 50 ml (0.05 moles) of a 1 M solution of calcium ethoxide in absolute ethanol.

After combining the calcium ethoxide and aluminum sec-butoxide, the mixture was stirred and allowed to stand for 48 hours. Needle shaped crystals were observed on the bottom of the reaction Vessel. The supernatant was decanted and the crystals were washed with absolute ethanol. Upon standing, the crystals were observed to disappear and the solution took on the consistency of a gel. Sufficient hydrochloric acid (1 N) was added to the gel to reduce the pH from 12 to 8.

Upon standing, the supernatant separated into a buff colored alcoholic layer and a cloudy colloidal layer. One ml of the alcoholic layer was added to 0.1 ml of 0.09 N hydrochloric acid to form a transparent gel. This gel was rinsed quickly with 2 ml of distilled water and dried in an oven at about 150°C. The resulting white powder was applied to carbon support grids for elemental analysis using a scanning and transmission electron microscope. Strong calcium and aluminum signals were present. It is believed that the gel prepared in this Example consists of an amorphous network of calcium aluminate ( $\text{CaAl}_2\text{O}_4$ ). Preliminary experiments indicate that flax paper dipped into the gels of this Example resist combustion.

#### EXAMPLE XI

##### Preparation Of Titania Gel

17.52 ml (0.3 moles) of absolute ethanol was added to a 50 ml beaker. The beaker was placed in an ice bath. A stream of nitrogen gas was directed into the beaker. While stirring vigorously with a magnetic stirrer, 11.16 ml (0.1 moles) of titanyl tetrachloride was added dropwise to the beaker. After the addition was complete, the beaker was removed from the ice bath. The mixture was stirred

for an additional 2 hours. At this time, the mixture was stored for later use.

#### EXAMPLE XII

##### Coating And Evaluation Of Papers With Gels Prepared By The Sol-Gel Process

Approximately 0.6 g of the gel of Example XI, which had solidified, was dissolved in 25 ml of 70% ethanol. Samples of commercial grade flax cigarette paper were coated by dipping as described in Example II. One sample was dipped once and allowed to dry. Another sample was dipped once, allowed to dry, dipped again, and allowed to dry. A third sample was dipped twice and then dried in a microwave oven as described in Example II. Scanning electron micrographs revealed that the gel of Example XI produced a uniform coating over the surface of the dipped papers.

#### EXAMPLE XIII

##### Preparation of Zirconia Gels

32.72 g (0.1 moles) of zirconium isopropoxide was dissolved in 200 ml of isopropanol. While stirring with a magnetic stirring hot plate, approximately 108.5 ml of isopropanol containing 8 ml of glacial acetic acid and 3.6 ml of distilled water was added. The mixture was then heated with constant stirring as in Example I. A gel formed within 5 minutes.

#### EXAMPLE XIV

##### Preparation Of Aluminum Oxide Gel With Hydrochloric Acid

15 ml of 1 N hydrochloric acid was added to 1500 ml of distilled water. The acid and water mixture was heated to 70°C. Approximately 120 g (0.5 moles) of aluminum sec-butoxide was then added. This mixture was heated to 95°C with stirring. The temperature of the solution was maintained at 95°C and stirring continued until all of the sec-butanol evaporated. This took approximately 2 hours. At this time, 30 ml (0.03 moles) of 1 N hydrochloric acid was added. The temperature was maintained and stirring continued until a final volume of 400 ml was obtained.

hand. Following application of the gel to paper, the gel is dried to form a film upon the paper. Preferably the gel is dried at a temperature of 100 to 150°C. The method of drying may be by any known method such as by contact with a heated surface.

When used as wrappers for smoking articles, wrappers of the present invention have superior burn characteristics when compared with untreated papers. These improvements include substantial reduction of sidestream smoke, reduced incidence of flaming combustion, and, if desired, complete elimination of combustion. Although not intending to be bound by theory, it is believed that during the combustion of gel-coated papers, the ceramics collapse to form a sheath around the paper and tobacco ash at the end of the smoking article. This sheath maintains its structural integrity because of the strength imparted by the cross-linked structure of the dried sol-gel film. The thickness, uniformity in coverage, and porosity of the sheath, in both the radial and longitudinal directions, determines the resulting burn characteristics of the smoking article.

A cigarette according to the invention can be manufactured according to methods well known in the art, except that a wrapper is formed from cigarette paper to which a thin and continuous coating of the gels of this invention has been applied. This coating may be applied to cigarette paper before cigarettes are manufactured by using, for example, size presses, rotogravure presses, or blade coaters. Use of cigarette paper coated according to this invention as wrappers for cigarettes reduces the amount of sidestream smoke given off during combustion compared with cigarettes made with untreated paper wrappers.

In another embodiment of this invention, an annular coating of the gel is applied to the paper fibers of the wrapper of a cigarette in at least one separate ring disposed perpendicularly to the longitudinal axis of a cigarette. This produces a cigarette that will cease burning if not actively smoked, i.e., drawn upon by the smoker, for a predetermined amount of time. Preferably, a plurality of discrete bands, each of which substantially circumscribes the periphery of a smoking article, are employed. This coating is preferably applied in the form of a gel which is subsequently dried to form a film upon the paper. The gel may be applied directly to cigarettes or it may be applied to cigarette paper, with a size press, rotogravure press, or a blade coater, which is then used to manufacture cigarettes. Such treatment results in a cigarette that will have a mass burn rate approaching zero if allowed to idle as when it is placed in an ashtray between puffs. The time required for a cigarette made according to this embodiment to approach zero mass burn rate varies with the location and

thickness of the film applied to the paper fibers.

In yet another embodiment of this invention, a smoking article of the type specified has a cigarette paper according to this invention. Alternatively, a 10 mm wide annulus of the gels of the present invention is applied about 8 mm from a distal or lighting end of such novel smoking articles. After the gel is dried, the resulting film protects the wrappers of these smoking articles from thermal damage including charring and flaming combustion.

As shown in Figure 1, preferred smoking articles of the type specified consist of a mouth end 13, a spacer element 12, and an active element 11 at a distal end remote from the mouth end. The active element 11 is in fluid communication with the mouth end 13. The active element 11 includes a heat reflective substantially hollow sleeve 22 having an internal wall 23 and an external wall 24 and having a first end at the distal end and a second end closer to the mouth end. A reflective end cap 15 may be clipped over the external wall 24 of the sleeve 22 at the first end of the sleeve. Cap 15 has one or more orifices or perforations 16 which allow air into active element 11. Disk 27 may be provided at the second end of sleeve 22 to close off active element 11 from spacer element 12 while still allowing fluid communication through orifice 28.

A substantially cylindrical carbon-containing heat source 20 is inserted in sleeve 22 adjacent the first end of the sleeve. Preferably, sleeve 22 is fitted with one or more metallic clips 17 which hold heat source 20 suspended away from internal wall 23 of sleeve 22, defining an annular space 25 around the heat source. The heat source 20 has a fluid passage 206 substantially through the center thereof.

A flavour bed 21 is held within sleeve 22 between clip 17 and heat source 20 on one end and a screen-like clip 26, which holds pellets of flavour bed 21 while allowing air passage, on the other end. Flavour bed 21 is in radiative and convective heat transfer relationship with heat source 20. Active element 11 and spacer element 12 are jointly wrapped in an abutting relationship by wrapper 14.

Mouth end 13 may include a filter segment 29 and a tobacco rod segment 30. Filter segment 29 may be a cellulose acetate filter plug 201 wrapped in plug wrap 202. Tobacco rod segment 30 may be tobacco filler 203 wrapped in plug wrap 208. Filter segment 29 and tobacco rod segment 30 are jointly overwrapped in an abutting relationship by plug wrap 204. Mouth end 13 is jointly overwrapped with spacer element 12, which is jointly overwrapped with active element 11, in an abutting relationship by tipping paper 205.

When the heat source is ignited and air is drawn through the smoking article, air is heated as

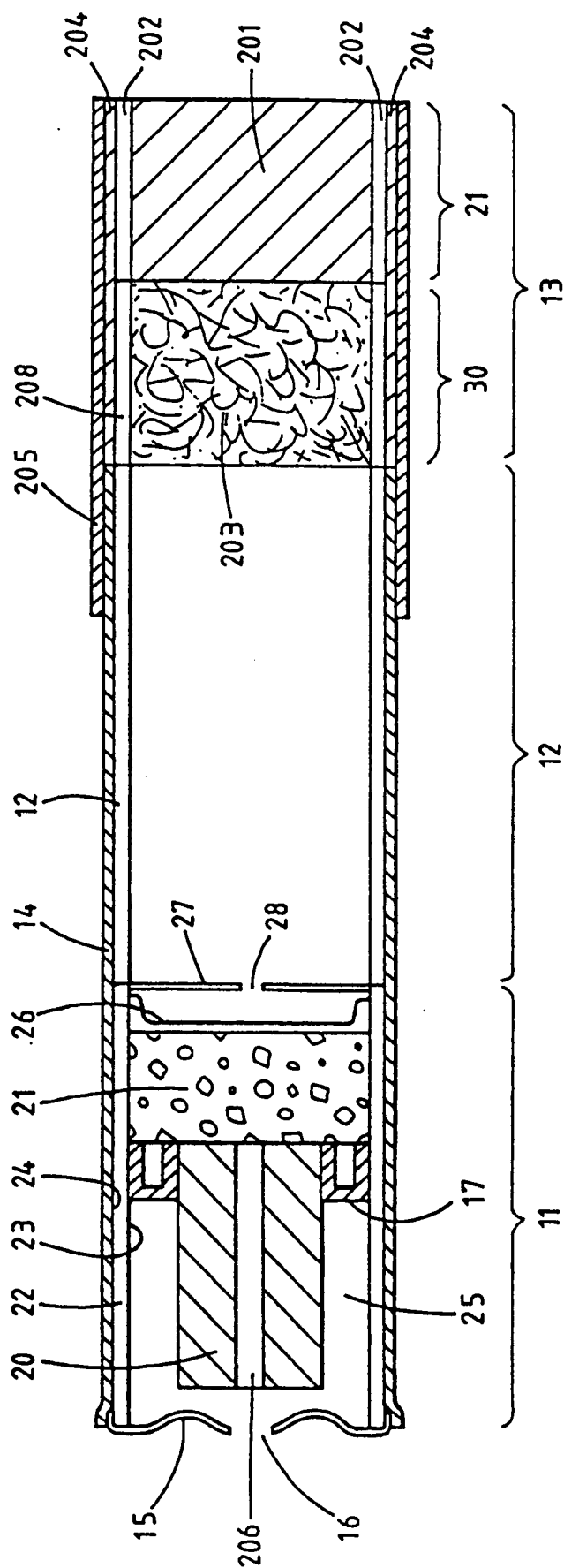


FIG. 1

## EXAMPLE XV

Preparation Of Aluminum Oxide Gel With Hydrochloric Acid

Approximately 240 g (1 mole) of aluminum-sec-butoxide was mixed with 1600 ml of water containing 30 ml (0.03 moles) of 1 N hydrochloric acid, which had been heated to 70°C prior to mixing. The mixture was brought rapidly to 95°C with continuous stirring and kept at that temperature for approximately 2.5 hours. At the end of that period of time, 60 ml of 1 N hydrochloric acid was added to the mixture. The temperature of the solution was maintained and stirring continued until a final Volume of 500 ml was obtained.

## EXAMPLE XVI

Reduction Of Mass Burn Rate

Conventional cigarettes were treated with gel prepared according to Example XIV. As shown in Figures 2 and 3, cigarette 100 was coated with an annulus of gel 101. The annulus of gel 101 was applied to the wrapper 102 of cigarette 100 by hand with a brush. The center of the annulus was placed 32.5 mm from the lighting end 103 of cigarette 100. One set of cigarettes was coated with an annulus 4 mm wide; another set of cigarettes was coated with an annulus 5 mm wide. The width of each annulus was measured parallel to the longitudinal axis of the cigarette. The coated cigarettes were dried in air until the gels formed films on the cigarettes.

Cigarettes from each set were then individually tested by lighting a cigarette and placing it on a pan balance. The change in the weight of the burning cigarette was measured over time. Figures 4 and 5 illustrate the change in the weight of one cigarette from each set against time. Figure 4 shows the result when a 4 mm band of gel was applied; Figure 5 shows the result when a 5 mm band of gel was applied.

The slope of the lines in Figures 4 and 5 represent the mass burn rate of the cigarette, that is, how much weight is lost during a given burn time. The negatively sloping mass burn rate beginning at time equal to 0 seconds, shown in Figures 4 and 5, illustrate that both cigarettes experience constant weight loss after lighting. However, the mass burn rate decreases when the burning coal reaches the film annulus on the cigarette wrapper. This point is marked as "A" in both figures. In Figure 4, this change occurred after approximately 197 seconds. In Figure 5, this change occurred after approximately 263 seconds. This change in slope illustrates that the annulus of film inhibits the

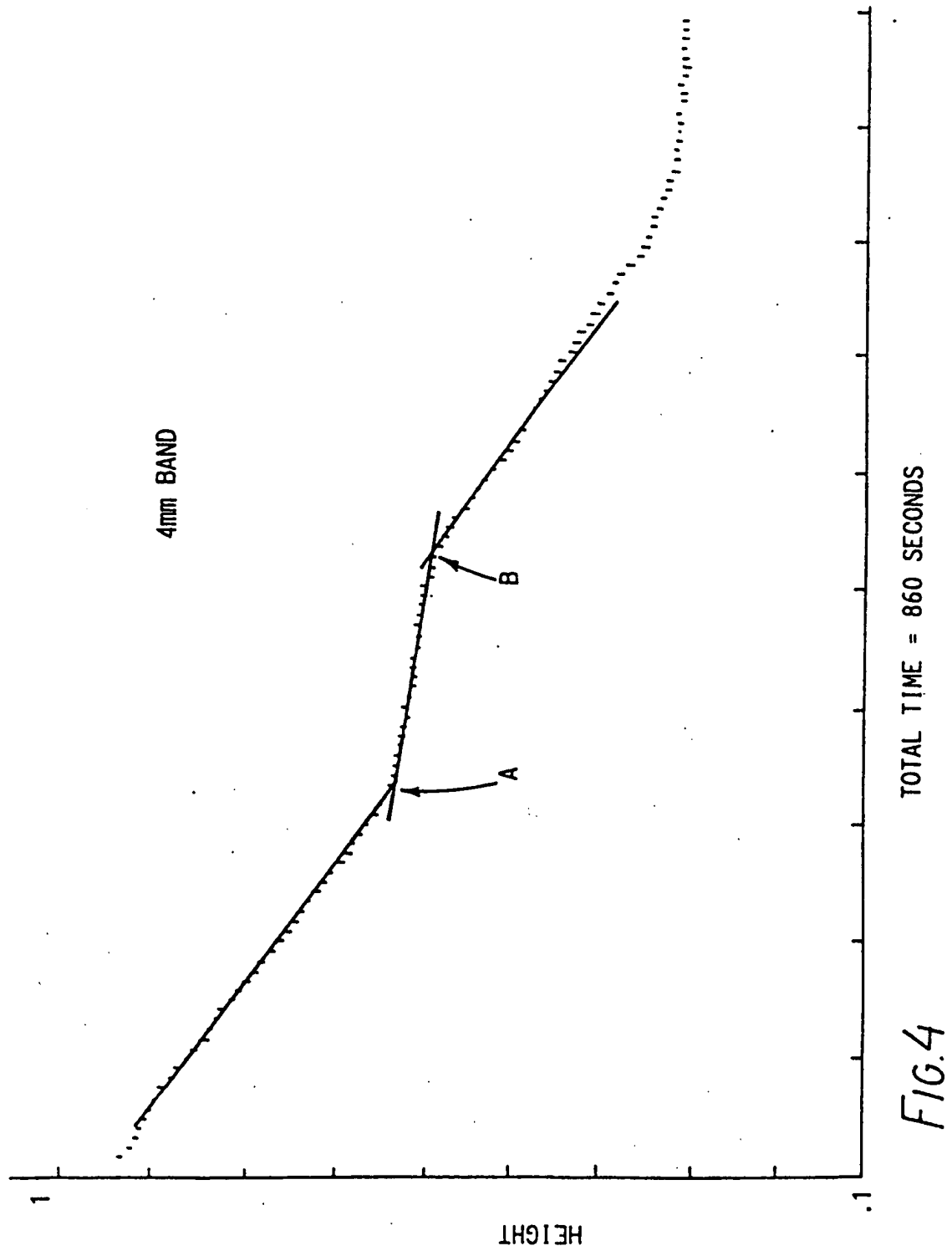
combustion of the cigarette.

As shown in Figure 4 at point "B" (at approximately 455 seconds), after the cigarette has burned through the annulus of film, the mass burn rate again increases. However, as shown in Figure 5 at point "C" (at approximately 417 seconds), the mass burn rate declines to zero indicating that the cigarette has extinguished. The affect of different widths of films is thus evident from a comparison of the mass burn rate lines. With a 4 mm wide film annulus, the cigarette remains lit after the burning coal has burned through the full width of the film annulus. However, with a 5 mm wide film annulus, the cigarette ceases burning after the burning coal reaches the film annulus but before burning through the annulus.

## Claims

1. A wrapper for a smoking article comprising paper fibres and at least one of aluminium oxide, zirconium oxide, titanium oxide, calcium oxide, magnesium aluminate and calcium aluminate.
2. A smoking article comprising a wrapper according to claim 1.





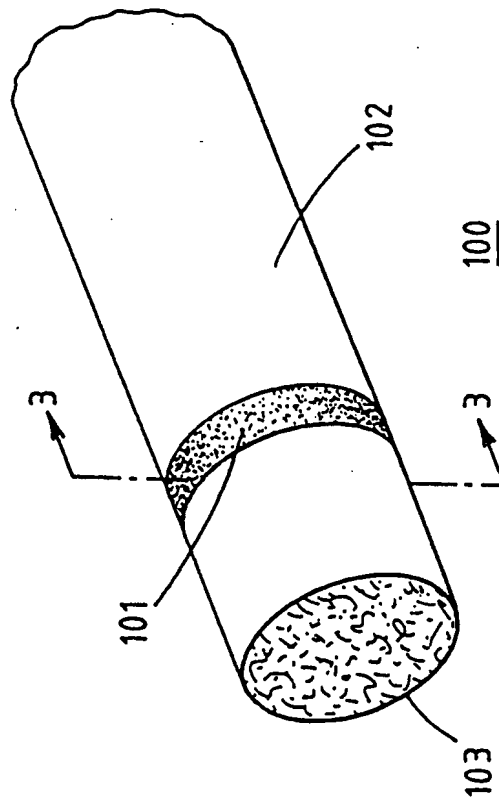


FIG. 2

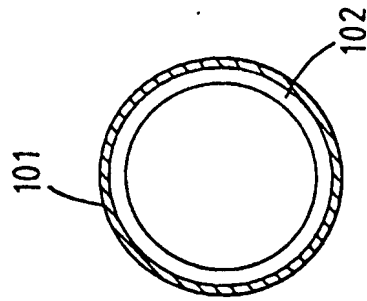
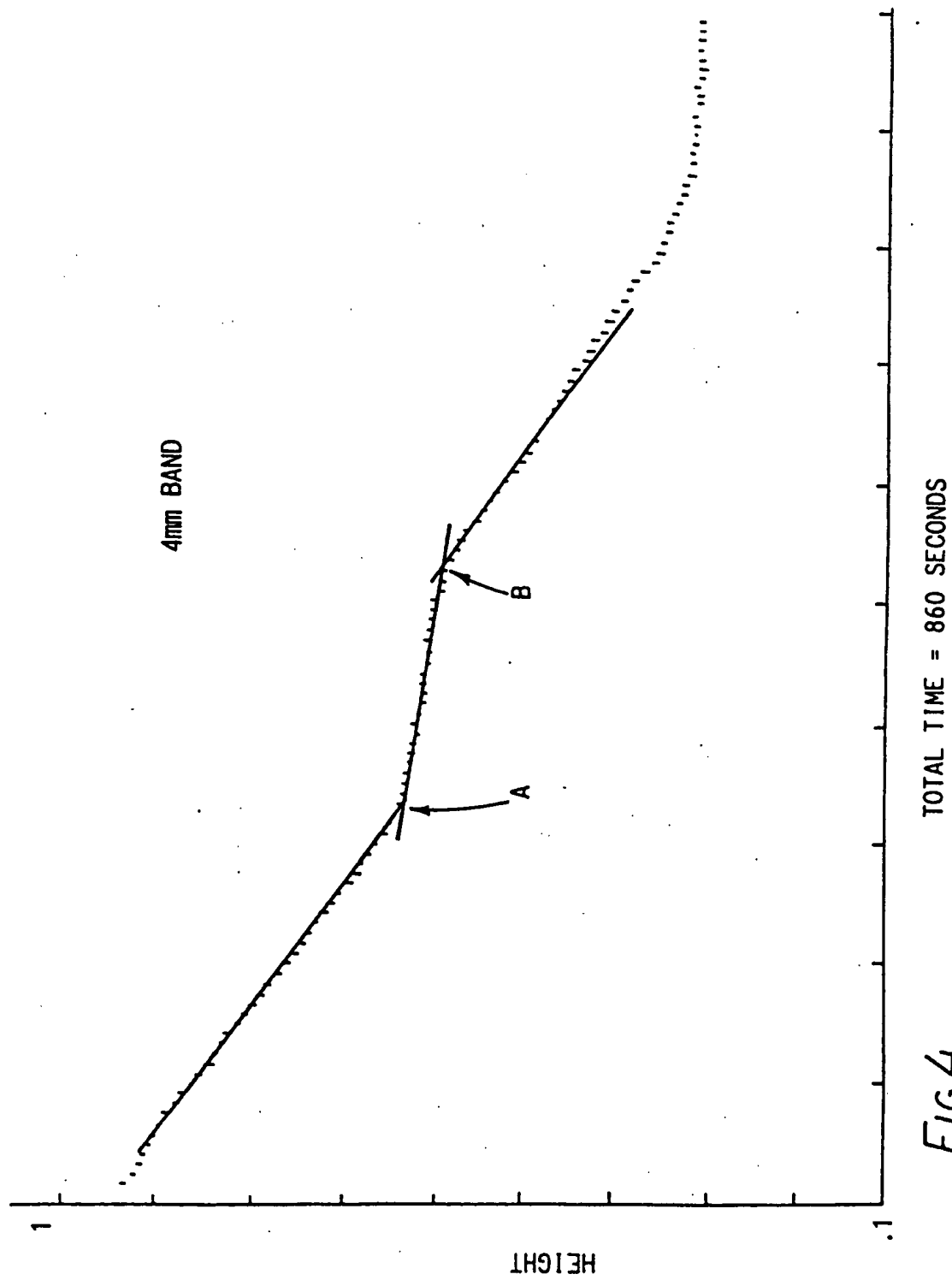


FIG. 3



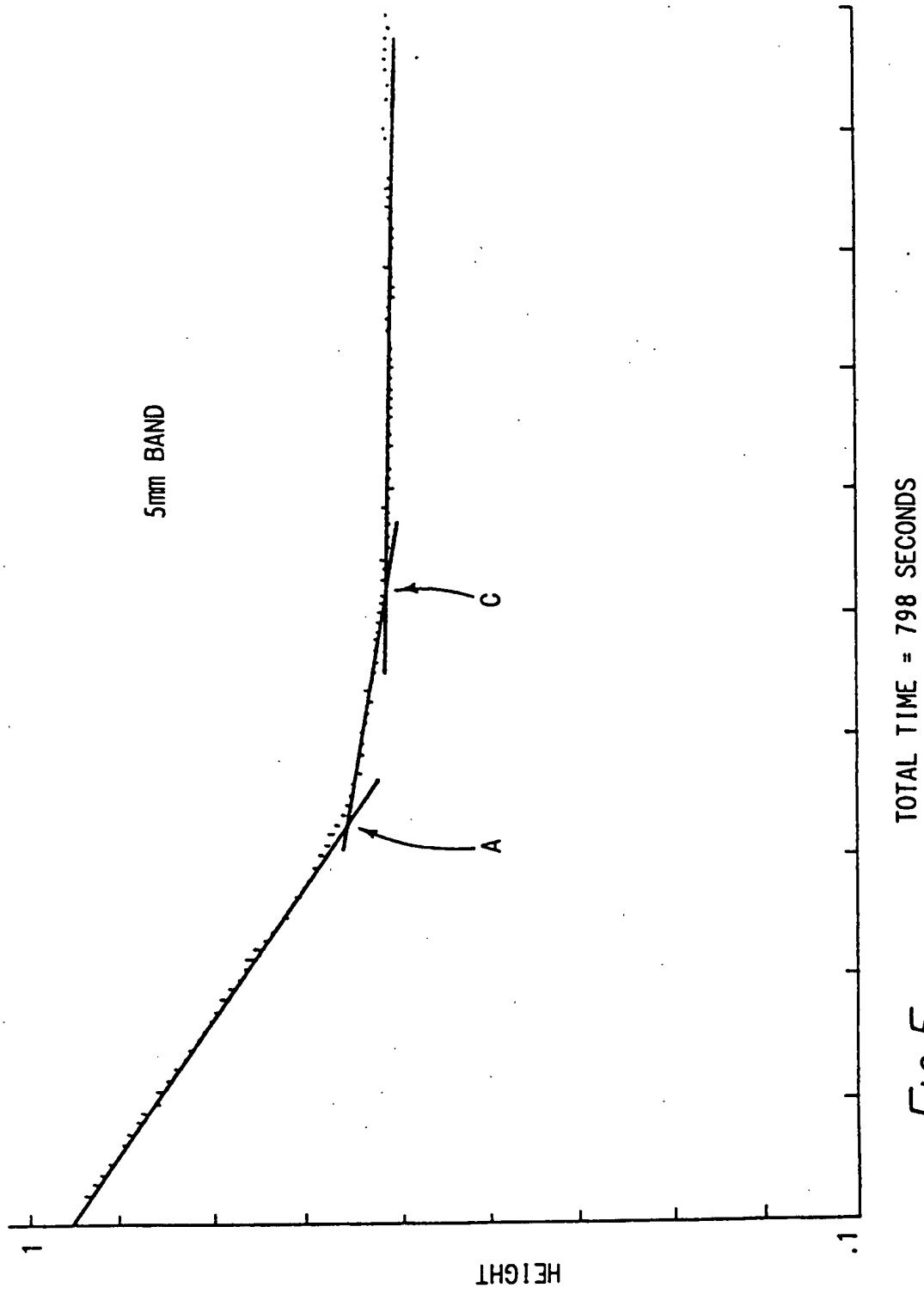


FIG.5